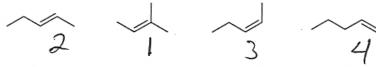
Chem 350 Test 3 Versione Answe

JASPERSE CHEM 350 TEST 3 Ch. 7 Structure and Synthesis of Alkenes

Ch. 8 Reactions of Alkenes

1. Rank the following alkenes in order of stability, 1 being most stable, 4 being least stable. (3 points)

VERSION 4

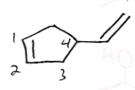


2. Determine the number of elements of unsaturation for C₅H₈O. (2 points)

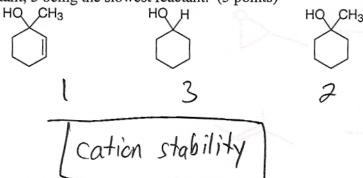
2

3. Give the proper IUPAC name or the structure for the following compounds. (3 points each)

a. 4-vinylcyclopentene



4. Rank the reactivity of the following alcohols towards H_2SO_4/Δ catalyzed dehydration, 1 being the fastest reactant, 3 being the slowest reactant. (3 points)



- 5. Which of the following statements is true for the structures shown: (3 points)
 - a. A is Z and B is Z
 - (b) A is Z and B is E
 - c. A is E and B is Z
 - d. A is E and B is E

$$A \stackrel{\text{CI}}{\longleftarrow} B \stackrel{\text{OCH}_3}{\longrightarrow}$$

6. Predict the <u>major</u> product for the following reactions. You needn't bother to show any side products or minor products. Pay careful attention to orientation, which is important in many of these problems. (3 points each)

e.
$$\bigvee_{Br} \frac{NaOCH_3}{EJ}$$

f.
$$\frac{\text{potassium}}{\text{t-butoxide}}$$
 $\frac{\text{t-butoxide}}{\text{(KOCMe}_3)}$
 $\frac{\text{balky base}}{\text{balky base}} \neq \text{E}_{2}$

7. Predict the <u>major</u> product in each of the following reactions. Pay careful attention to stereochemistry: stereochemistry is involved in each of these problems! (3 points each)

- b. OsO₄ OHOH
- c. 1. BH₃•THF
 2. H₂O₂, NaOH
- d. CH3CO3H, H2O
- 8. Fill in the intermediates in the following transformation. (3 points each)

OH
$$\frac{\text{H}_2\text{SO}_4, \Delta}{\text{A}}$$
 $\frac{\text{HBr,}}{\text{Peroxides}}$
 $\frac{\text{B}}{\text{NEt}_3, \Delta}$

OMe $\frac{1. \text{Hg(OAc)}_2, \text{MeOH}}{2. \text{NaBH}_4}$

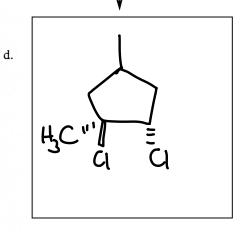
C

9. For the following reactions, fill in the missing <u>starting materials</u>, <u>reagents</u>, <u>or products</u>. (3 points each)

c.
$$\frac{1. O_3}{2. Me_2S}$$

$$\frac{1. O_3}{2. Me_2S}$$

$$\frac{1. O_3}{2. Me_2S}$$



- 10. Provide the major product of the following reaction sequences. (4 points each)
- 1. H₂O, H⁺
 2. H₂SO₄, Δ

- 1. HBr, peroxides
 2. KOCMe₃
 3. Br₂
- 11. Provide reagents to accomplish the following transformations. (4 points each)

$$OH \frac{1. H_2SC_{4,} H^{\dagger}}{2. O_SC_{4,} H_2C_2} \xrightarrow{CH_3}$$

12. Provide the product for the following reaction. Be sure to show the stereochemistry of the product. (3 points)

13. Draw the mechanism for the following reaction, and <u>write "slow" next to the rate determining step</u>. Be sure to draw all intermediates, and to correctly draw "electron-movement" arrows or half-arrows. (4 points)

14. Draw the mechanism for the following reaction. Be sure to draw all intermediates, and to correctly draw "electron-movement" arrows or half-arrows. (4 points)

$$\frac{Br_2}{\int J_3 r - J_3 r} Br$$

15. Formula: C_4H_8 $\mathcal{E}\mathcal{U}=$

Reactivity: reacts with H2/Pt to give C4H10 alkere

DRAW ALL POSSIBLE ISOMERS, INCLUDING STEREOISOMERS. (4 isomers are possible!) (5 points)