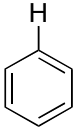
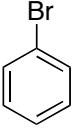
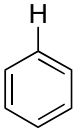
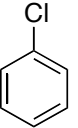
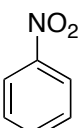




5 Major Electrophilic Aromatic Substitution Reactions

		Activating/ Deactivating	Ortho/Para Or Meta Directing	Book
1	 + Br ₂ $\xrightarrow[\text{(or Fe cat.)}]{\text{FeBr}_3 \text{ (cat.)}}$  (+ HBr)	Deactivating	Ortho/Para	17.2
	 + Cl ₂ $\xrightarrow{\text{AlCl}_3 \text{ (cat.)}}$  (+ HCl)	Deactivating	Ortho/Para	17.2

The halides are unique in being deactivating but ortho/para directing. All other o/p-directors are activating, and all other deactivating groups are m-directors. **Mech**

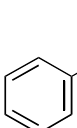

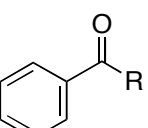
2	 + HNO ₃ $\xrightarrow{\text{H}_2\text{SO}_4}$  (+ H ₂ O)	Deactivating	Meta	17.3
---	---	--------------	------	------

The product can be reduced to Ar-NH₂ by Fe/HCl or Sn/HCl. Nitration/Reduction provides an effective way to introduce an NH₂ group. Reduction converts m-directing NO₂ group into an o/p-directing NH₂ group. **Mech required.**

3	 + R-X $\xrightarrow{\text{AlCl}_3 \text{ (cat.)}}$  (+ HCl)	<u>Activating</u>	Ortho/para	17.10
---	--	--------------------------	------------	-------


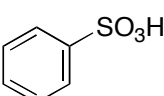
- Restricted to 3°, 2°, or ethyl halides. 1° halides suffer carbocation rearrangements.
- Since product is more active than starting material, polyalkylation is often a serious problem.
- Fails with strongly deactivated benzenes.

Mech required.

4	 +  $\xrightarrow{\text{AlCl}_3 \text{ (cat.)}}$  (+ HCl)	Deactivating	Meta	17.11
---	--	--------------	------	-------

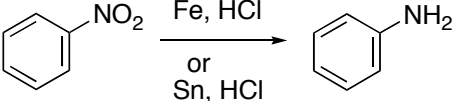
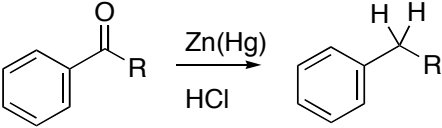
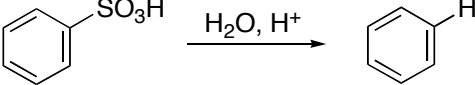
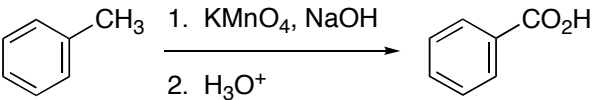
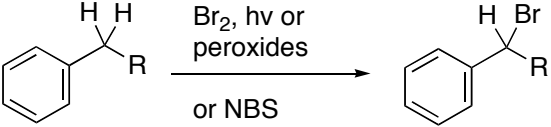
- The product can be reduced to -CH₂R by Zn(Hg)/HCl.
- The acylation-reduction sequence provides an effective way to introduce a 1° alkyl group.
- Reduction converts meta-directing acyl group into an ortho/para-directing alkyl group.

Mech required.

5	 + SO ₃ $\xrightarrow{\text{H}_2\text{SO}_4}$ 	Deactivating	Meta	17.4
---	---	--------------	------	------

The sulfonyl group is a useful para-blocking group, since it can later be removed upon treatment with H₂O/H⁺. **No mech required.**

5 Major Aromatic Support Reactions

			<u>Activating/ Deactivating</u>	<u>Ortho/Para Or Meta Directing</u>	<u>Book</u>
6			Activating	Ortho/Para	19.21
	<ul style="list-style-type: none"> Reduction converts meta-director into an ortho-para director. Fe, Sn, or several other reducing metals can work. No mech required. 				
7			Activating	Ortho/Para	17.12
	<ul style="list-style-type: none"> Clemmensen reduction converts meta-director into an ortho-para director. Acylation (#4) followed by Clemmensen Reduction (#7) is the standard method for introducing a 1° alkyl group. (Direct alkylation with a 1° alkyl halide, reaction #3, fails due to cation rearrangement problems...) No mech required. 				
8		-----	-----		17.4
	<ul style="list-style-type: none"> The sulfonyl group is a useful and reversible para-blocking group, since it can be temporarily put on (reaction 5) but then can be removed later upon treatment with H₂O/H⁺ (reaction 8). The sulfonation/other reaction/desulfonation sequence is crucial for clean ortho-substitution of an o/p director. 				
	No mech required.				
9			Deactivating	Meta	17.14
	<ul style="list-style-type: none"> Oxidation converts ortho/para-director into a meta-director. Side alkyl chains longer than methyl can also be oxidized to benzoic acid in the same way, although more time and heat is required. For test purposes, just writing KMnO₄ will be OK. But the real reaction requires a basic solution for the KMnO₄ to work, so an acidic workup step is actually required to isolate the neutral carboxylic acid. 				
	No mech required.				
10		-----	-----		17.14
	<ul style="list-style-type: none"> Bromination occurs via free-radical mechanism. It is selective for substitution at the benzylic position because the benzylic radical intermediate is resonance-stabilized. Note: keep distinct Br₂/FeBr₃ from Br₂/peroxides! Product is subject to S_N2 substitutions (benzylic bromides are especially good, better than normal 2° bromides) and E2 eliminations with bulky bases. "NBS" is N-bromosuccinimide, which functions just like Br₂/peroxides, but is much more convenient and cleaner because it avoids competing reactions caused by lots of Br₂ and HBr. 				
	<u>Mech required.</u>				