

## 5 Major Electrophilic Aromatic Substitution Reactions

		<b>Activating/ Deactivating</b>	<b>Ortho/Para Or Meta Directing</b>	<b>Book</b>
1		Deactivating	Ortho/Para	17.2
		Deactivating	Ortho/Para	17.2

The halides are unique in being deactivating but ortho/para directing. All other o/p-directors are activating, and all other deactivating groups are m-directors. **Mech**

2		Deactivating	Meta	17.3
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The product can be reduced to Ar-NH<sub>2</sub> by Fe/HCl or Sn/HCl. Nitration/Reduction provides an effective way to introduce an NH<sub>2</sub> group. Reduction converts m-directing NO<sub>2</sub> group into an o/p-directing NH<sub>2</sub> group. **Mech required.**

3		<b><u>Activating</u></b>	Ortho/para	17.10
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- Restricted to 3°, 2°, or ethyl halides. 1° halides suffer carbocation rearrangements.
- Since product is more active than starting material, polyalkylation is often a serious problem.
- Fails with strongly deactivated benzenes.

**Mech required.**

4		Deactivating	Meta	17.11
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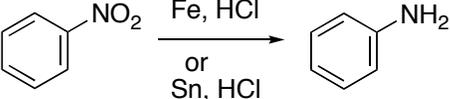
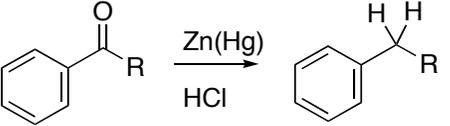
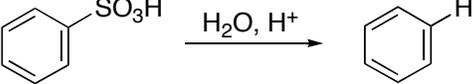
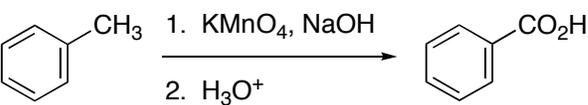
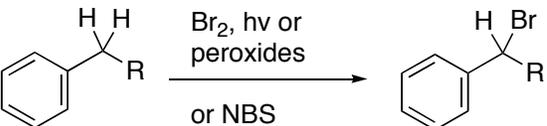
- The product can be reduced to -CH<sub>2</sub>R by Zn(Hg)/HCl.
- The acylation-reduction sequence provides an effective way to introduce a 1° alkyl group.
- Reduction converts meta-directing acyl group into an ortho/para-directing alkyl group.

**Mech required.**

5		Deactivating	Meta	17.4
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The sulfonyl group is a useful para-blocking group, since it can later be removed upon treatment with H<sub>2</sub>O/H<sup>+</sup>. **No mech required.**

**5 Major Aromatic Support Reactions**

		<u>Activating/ Deactivating</u>	<u>Ortho/Para Or Meta Directing</u>	<u>Book</u>
6		Activating	Ortho/Para	19.21
	<ul style="list-style-type: none"> <li>Reduction converts meta-director into an ortho-para director.</li> <li>Fe, Sn, or several other reducing metals can work. No mech required.</li> </ul>			
7		Activating	Ortho/Para	17.12
	<ul style="list-style-type: none"> <li>Clemmensen reduction converts meta-director into an ortho-para director.</li> <li>Acylation (#4) followed by Clemmensen Reduction (#7) is the standard method for introducing a 1° alkyl group. (Direct alkylation with a 1° alkyl halide, reaction #3, fails due to cation rearrangement problems...) No mech required.</li> </ul>			
8		-----	-----	17.4
	<ul style="list-style-type: none"> <li>The sulfonyl group is a useful and reversible para-blocking group, since it can be temporarily put on (reaction 5) but then can be removed later upon treatment with H<sub>2</sub>O/H<sup>+</sup> (reaction 8).</li> <li>The sulfonation/other reaction/desulfonation sequence is crucial for clean ortho-substitution of an o/p director.</li> </ul> <p style="text-align: right;">No mech required.</p>			
9		Deactivating	Meta	17.14
	<ul style="list-style-type: none"> <li>Oxidation converts ortho/para-director into a meta-director.</li> <li>Side alkyl chains longer than methyl can also be oxidized to benzoic acid in the same way, although more time and heat is required.</li> <li>For test purposes, just writing KMnO<sub>4</sub> will be OK. But the real reaction requires a basic solution for the KMnO<sub>4</sub> to work, so an acidic workup step is actually required to isolate the neutral carboxylic acid.</li> </ul> <p style="text-align: right;">No mech required.</p>			
10		-----	-----	17.14
	<ul style="list-style-type: none"> <li>Bromination occurs via free-radical mechanism.</li> <li>It is selective for substitution at the benzylic position because the benzylic radical intermediate is resonance-stabilized.</li> <li>Note: keep distinct Br<sub>2</sub>/FeBr<sub>3</sub> from Br<sub>2</sub>/peroxides!</li> <li>Product is subject to S<sub>N</sub>2 substitutions (benzylic bromides are especially good, better than normal 2° bromides) and E2 eliminations with bulky bases.</li> <li>“NBS” is N-bromosuccinimide, which functions just like Br<sub>2</sub>/peroxides, but is much more convenient and cleaner because it avoids competing reactions caused by lots of Br<sub>2</sub> and HBr.</li> </ul> <p style="text-align: right;"><b><u>Mech required.</u></b></p>			