1. Provide names or structures for the following. 2 points each. Specify stereochemistry when appropriate!

- Cl
- CH₃

Br

m-isopropylphenol

Cl

HO

2. Identify the functional groups in the following molecule. (4 points)
3. Predict the major products for the following reactions. Pay careful attention when orientation is a factor. Draw just one major product in each case. (3 points each)

\[ \text{OCH}_3 + \text{CN} \rightarrow \]

\[ \text{Br}_2, \text{hv} \]

\[ \text{Br} \]

\[ \text{O}_3 \]

\[ \text{Me}_2\text{S} \]

\[ \text{HBr} \]

\[ \text{OsO}_4, \text{H}_2\text{O}_2 \]

\[ \text{Br}_2, \text{FeBr}_3 \]

\[ \text{Zn(Hg), HCl} \]

\[ \text{Br}_2, \text{hv} \]

\[ \text{H}_2\text{O}, \text{Hg(OAc)}_2 \]

\[ \text{NaBH}_4 \]
4. Classify the pairs of molecules as totally different, identical, structural isomers, diastereomers, or enantiomers. (2 points each)

5. Classify each chiral carbon as R or S. (2 points each)

6. Classify the hybridization and bond angles (109, 120, or 180) at the labelled atoms. (5 points)
7. Draw the mechanisms for the following reactions. For any radical reactions, draw propagation steps only. 5 points each.

\[ \text{OCH}_3 \quad \text{Cl}_2 \quad \text{AlCl}_3 \quad \text{OCH}_3 \]

Draw the resonance structures for the cation intermediate.

\[ \text{Ph} \quad \text{H} \quad \text{H} \quad \text{Ph} \quad \text{H} \quad \text{Br} \quad \text{Br} \quad \text{H} \quad \text{Br} \quad \text{H} \quad \text{Br} \quad \text{Ph} \quad \text{Ph} \]
8. Draw the products of the following multi-step sequences. (4 points each)

\[
\begin{align*}
\text{Me} & \quad \text{Br} \\
\text{Ph} & \\
\text{H} \quad \text{NaOH} \\
\text{HBr, peroxides} \\
\text{NaOMe}
\end{align*}
\]

\[
\begin{align*}
\text{CH}_3 \\
\text{1. SO}_3, \text{H}_2\text{SO}_4 \\
\text{2. 2-bromopropane, AlCl}_3 \\
\text{3. H}_2\text{O, H}^+
\end{align*}
\]

\[
\begin{align*}
\text{OH} \\
\text{1. H}_2\text{SO}_4 \\
\text{2. HBr, Peroxides} \\
\text{3. NEt}_3
\end{align*}
\]

9. Draw as many structural isomers as you can for C\textsubscript{6}H\textsubscript{14}. Circle any that are chiral. (Note: be careful! You will lose points for any repeats!) (6 points)
10. Rank the Following, from most to least. 2 points each.

a. Reactivity toward $S_{N2}$

\[ \text{Br} \quad \text{Cl} \quad \text{I} \quad \text{Cl} \]

b. Stability

\[ \text{Me}_2\text{CH} \quad \text{H} \quad \text{CH}_3 \quad \text{H} \]

c. Stability

\[ \text{Me}_2\text{CH} \quad \text{H} \quad \text{Me}_2\text{CH} \quad \text{H} \]

d. Acidity

\[ \text{PhCO}_2\text{H} \quad (\text{CH}_3)_2\text{NH} \quad \text{HCl} \quad \text{CH}_3\text{CH}_2\text{OH} \]

e. Stability

\[ \text{Me} \quad \text{Me} \quad \text{Me} \quad \text{Me} \]

f. Reactivity toward $\text{H}_2\text{SO}_4$/$\text{I}$ catalyzed dehydration

\[ \text{OH} \quad \text{OH} \quad \text{OH} \]

\[ \text{OH} \quad \text{OH} \quad \text{OH} \]

\[ \text{OH} \quad \text{OH} \quad \text{OH} \]

g. Boiling Point

\[ \text{Me}_2\text{CH} \quad \text{H} \quad \text{Me}_2\text{CH} \quad \text{OH} \]

\[ \text{Me}_2\text{CH} \quad \text{H} \quad \text{Me}_2\text{CH} \quad \text{OH} \]

\[ \text{Me}_2\text{CH} \quad \text{H} \quad \text{Me}_2\text{CH} \quad \text{OH} \]
11. Provide reagents for the following transformations. You may use anything you like. Each can be done within $\leq 3$ steps. (4 points each)

\[ \text{Ph - CH}_3 \text{CH}_3 \rightarrow \text{Ph - OCH}_3 - \text{CH}_2 - \text{CH}_3 \]

\[ \text{C}_6\text{H}_{11} \text{OH} \rightarrow \text{C}_6\text{H}_{11} \text{Br} \]

\[ \text{Ph - CH}_3 \text{CH}_3 \rightarrow \text{Ph - CH}_3 \text{CH}_3 - \text{N}_2 \text{H} - \text{Br} \]

12. Provide the appropriate reactant for the following transformation. (3 points)

\[ \text{CH}_3\text{CO}_2\text{H}, \text{H}_2\text{O} \rightarrow \text{HOH} - \text{HOH} \]
13. Suggest a structure for X, given the following info: (5 points)

- **Formula:** \( \text{C}_8\text{H}_{12} \)
- It Reacts With excess \( \text{H}_2/\text{Pt} \) to produce \( \text{C}_8\text{H}_{16} \)
- When it reacts with \( \text{O}_3/\text{Me}_2\text{S} \), one of the products is \( \text{CH}_2=\text{O} \).

14. Which of the following are aromatic

![Structures](image)

15. Draw the products and mechanism for the following reaction:

![Reaction](image)

16. Rank the following:

![Reactivity](image)