JASPERSE

CHEM 342

TEST 4

VERSION 3

Ch 24 Amines

Ch 20 Carboxylic Acids

Ch 21 Carboxylic Acid Derivatives

- 1. Nomenclature. Provide Either the Name or the Structure for the Following Chemicals. (10 points)
 - a. N-ethyl-N-methyl-4-methyl-1-pentanamine

N-ethyl-N-methyl-4-methylpentan-1-amine

b. sodium (R)-3-hydroxybutanoate

4-methylpentanoyl chloride

cis-2,2-dimethyl-4-hexenoic acid
(Z) (Z)-2,2-dimethylhex-4-enoic acid

он (A)-4-amino-5-methy l hexanoic acid

ethyl pentanoate

Oops! Ethyl Butanoate, not pentanoate

2. For each nitrogen a-f, identify the hybridization of the nitrogen atom, and identify the hybridization of the nitrogen lone pair. [Adenine is an important player in information transfer (DNA, RNA, genetics, etc.) and energy storage/release (ATP/ADP).]

ŀ	Le C	√NH ₂
a _N	\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	
H, p	N.C.	

Nitrogen Atom	Hybridization of the Nitrogen Atom	Hybridization of the Nitrogen Lone Pair
<u>a</u>	Sp2	Sf ²
<u>b</u>	Sp ²	P
<u>c</u>	Sp2	SP2
<u>d</u>	592	Spz
<u>e</u>	Sp2	P
<u>f</u>	5p2 5p2 5p2 5p3	sp 3
	. 7	ÿ

3. Synthesis Reactions. Draw the feature product of the following reactions (need not show any byproducts). (15 points)

a. Ph
$$\longrightarrow$$
 Br $\xrightarrow{1. \text{ Mg}}$ 2. CO_2

c.
$$\frac{0}{2. \text{ H}_3\text{O}^+}$$

$$\frac{1. \text{ LiAlH}_4}{2. \text{ H}_3\text{O}^+}$$

$$\frac{0}{1. \text{ LiAlH}_4}$$

$$\frac{1. \text{ LiAlH}_4}{0 \text{ H}} = \frac{1. \text{ HC}}{0 \text{ HC}}$$

d.
$$\begin{array}{c} \begin{array}{c} \begin{array}{c} 1. \text{ SOCl}_2 \\ 2. \text{ Me}_2\text{NH (excess)} \end{array} \end{array} \\ \hline 3. \text{ LiAlH}_4; \text{H}_2\text{O} \end{array}$$

f. Ph OCH₃
$$\xrightarrow{1. \text{ NaOH, H}_2\text{O}}$$
 $\xrightarrow{\text{Ph}}$ OCH₃

4. Synthesis Reactions. Draw the feature product of the following reactions (need not show any byproducts). (15 points)

b.
$$\frac{1. \text{ HNO}_3, \text{H}_2\text{SO}_4}{2. \text{ Br}_2 \text{ FeBr}_3}$$

e.
$$\frac{1. \text{ KMnO}_4}{2. \text{ excess MeLi}}$$

$$\frac{1. \text{ KMnO}_4}{3. \text{ H}_3\text{O}^+}$$

Oops! These answers are each short one carbon!

Note is a spectator counterion. Needs no consideration in mech.

addn product

6. Provide Reagents for the following Transformations (15 points)

- 7. Which, when dissolved in diethyl ether, will: (5 points each)
- a) Extract into NaOH/H2O?

A+D

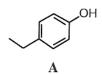
(veschance)

b) Extract into HCl/H2O?

B

c) Extract into water?

None



Me₂N B

HO C

8. Hydrolysis Reactions. Draw the starting materials for the following hydrolysis reactions. (6 points)

1. NaOH, H₂O

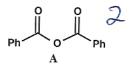
2. HCl

CO₂H + HO

Ph

1. NaOH, H₂O 2. HCl

9. Rank the following according to their reactivity toward NaOH/ H_2O hydrolysis.



Ph NHCH₃

Ph C OCH₃



Given the structures **A-D** above, which of the following reactions will proceed spontaneously? (2 points)

 $A + H_2NCH_3 \rightarrow B$

 $A + HOCH_3 \rightarrow C$

 $A + HCl \rightarrow D$

Yes Yes No

- 10. Rank the acidity of the following, 1 being most acidic, 3 being least (3 points each)
- a. acetic acid vs. water vs. NH_4 +Cl-
- b. CH_3OH vs. CH_3NH_2 vs. F_2CHOH
- c. p-methoxybenzoic acid vs. benzoic acid vs. acetone
- 11. Rank the basicity of the following, 1 being most basic, 3 being least (3 points each)
- a. CH_3OH vs. $PhNH_2$ vs. CH_3NH_2
- b. O_2N V_S . V_S .
- c. $\frac{1}{2}$ (CH₃CH₂)₃N H₂C $\frac{1}{3}$