

# ORGANIC CHEMISTRY I: CHEMISTRY 350 SYLLABUS

## FALL 2010

Classroom: SL118

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Office Hours:

M, W, F 11-1

T, H 1:30-3:30

### Required Text and Materials:

1) Text: "Organic Chemistry", 6th edition, by Wade (Note: if you have a different Wade edition, or a version of McMurry's Organic Chemistry as used at NDSU, contact me in order to use what you already have.)

2) Solutions Manual: "Solutions Manual, Organic Chemistry, 6th Edition", by Simek, Wade

Note: These aren't the newest versions, so you can buy used ones cheap on-line.

### Test Schedule

Test #1: Friday, Sept. 17	Ch. 1 Introduction and Review Ch. 2 Structure and Properties of Organic Molecules Ch. 3 Structure and Stereochemistry of Alkanes
Test #2: Wednesday, Oct. 20	Ch. 4 The Study of Chemical Reactions Ch. 5 Stereochemistry Ch. 6 Alkyl Halides: Nucleophilic Substitution and Elimination
Test #3: Wednesday, Nov. 10	Ch. 7 Structure and Synthesis of Alkenes Ch. 8 Reactions of Alkenes
Test #4: Friday, Dec. 3	Ch. 10 Structure and Synthesis of Alcohols Ch. 11 Reactions of Alcohols
Final Exam Thursday, December 9, 9:00	Comprehensive Final Exam

### Grading Summary:

Tests	400 points (4 x 100)	<u>Tentative letter grades</u>	A	90%
Final exam	150 points (1 x 150)		B	78%
Take-Home Quizzes	60-80 points? (6-10 of these?)		C	66%
+10 points extra credit possible for perfect attendance			D	54%

**THE INSTRUCTOR MAY LOWER BUT WILL NOT RAISE THE PERCENTAGE REQUIRED FOR A LETTER GRADE**

Jasperse website: <http://www.mnstate.edu/jasperse/>

This will include copies of:

- Handouts/notes/in-class practice problems
- quizzes
- practice tests and practice text answers

Final Exam: The final exam will be **cumulative**, covering all of the same material tested previously on Tests 1-4.

Take-Home Quizzes: I will assign a number of take-home "quizzes" (maybe 7-10 over the course of the semester?) These will normally be given out at least four days before they are due.

Homework and Study Strategy: All assigned book problems represent what I consider to be reasonable test-level problems. All have worked-out answers in the Solutions Manual. **The homework is a great way to practice problem solving, assess your progress, and prepare for tests.** Since solutions are available, I will not collect the book homework. **The few take-home quiz problems that I collect and grade are no substitute for doing book homework problems!**

Note: Putting off the extensive information in organic chemistry till the week of a test will only make it harder on you. After each class, try to work all of the assigned book problems at the back of the sections covered so that you will not only understand what you are doing at the time, but will remember how to do it weeks later! After each chapter is completed in class, do all of the assigned end-of-chapter problems in the book.

- A copy of the solutions manual is on reserve in the library, CHE-195.
- A copy of the textbook is on reserve in the library, CHE-207.

Attendance: Faithful attendance is important (and I do care if you come!) To reinforce your self-discipline, perfect attendance will be rewarded with 10 points of extra credit and a single absence with 5 points of extra credit. I will excuse absences at my discretion if the reason is good (illness or school events) and I am notified. Be sure to sign the attendance sheet each day!

### **Class E-Mail List**

An email list will be used to notify you of special scheduling information or other miscellany. (If I am sick and won't be able to hold class; when and where practice tests are to be held; if there are errors in one of the practice tests or book problems or in something I communicated in class, etc.) The list defaults to your official mnstate e-mail address. If that isn't what you actually use, contact me.

- Note: A test e-mail has already been sent. If you haven't received one, something needs correction.

### **Course Description**

CHEM 350 Survey of Organic Chemistry: Part I (3 credits)

Introduction to the classification, structure, reactions, and reaction mechanisms of carbon compounds.

**Prerequisite:** Chem 210

- Note: Organic Chemistry Laboratory I, Chem 355, is a related but separate class. It is not required, but if you want to be in the lab you must be registered for it.

### **Student Learning Outcomes/Course Objectives**

The general outcome goals are that students will understand the classification, structure, nomenclature, reactions, reaction mechanisms, and synthesis of carbon compounds including halocarbons, alkenes, and alcohols. A general summary of major learning topics is summarized on page 1, with the listing of chapters that will be covered. A more detailed list of learning topics is summarized on page 5, with an approximately day-by-day listing of topic coverage. Most of the learning outcomes will be assessed by problems in which students must demonstrate their understanding. The list of problems on page 3 represents a detailed and representative sampling of the types of problems that should be solvable by a student who has achieved the learning outcomes.

For a correlation of class coverage of **NCATE standards**, see the NCATE SYLLABUS APPENDIX on page 5.

### **Academic Honesty**

Cheating will not be tolerated and will be reported to the Dean of your College and the Vice President for Academic Affairs. It may also be reported to the Student Conduct Committee for further disciplinary action. For a full description of the MSUM Academic Honesty Policy, please see the Student Handbook. (<http://www.mnstate.edu/sthandbook/POLICY/index.htm>)

### **Special Accommodations**

Students with disabilities who believe they may need an accommodation in this class are encouraged to contact Greg Toutges, Coordinator of Disability Services at 477-5859 (Voice) or 1-800-627-3529 (MRS/TTY), CMU 114 as soon as possible to ensure that accommodations are implemented in a timely fashion.

### **Textbook website:**

[http://wps.prenhall.com/esm\\_organic\\_wade\\_6](http://wps.prenhall.com/esm_organic_wade_6) The textbook website has additional practice problems, practice exams, visualization helps, and links to other useful web-based study resources. But **you have to pay for access.**

## CHEMISTRY 350 PROBLEMS, Based on Wade Version 6

FALL 2010 Dr. Craig P. Jasperse (note: if you have a different Wade version or a McMurry, I can work with you to use your existing book rather than spending for a new one.)

Ch	In Chapter Problems	End of Chapter Problems
1	1(Si only), 2a-f, 3a-g, 4, 5a-c, 6(all!), 7a,b,d,e,g, 8a,e,f,g,h, 9, 10d-h, 11, 15, 17a, 18a-c, 19a-f [determine which is the "nucleophile" (electron pair donor) and which is the "electrophile" (electron pair receiver), and draw the arrows to show bond making and breaking. Do not do the "Bonsted-Lowry" discussion.]	21, 23, 25-29, 31, 32, 34-37, 40-43, (for 42 and 43, you should be able to process H <sub>2</sub> SO <sub>4</sub> by memory, the others by structure without needing to look at a list of acidity values), 44 (use nucleophile/electrophile designation, and definitely practice the arrow pushing), 46
2	1b (draw), 2 (skip part about 104.5° angle as opposed to 109° angle), 3, 4, 5a-f, 7a,b, 8, 9, 10 (three do, three don't; beware of "e", which is deceptive), 11, 16, 17 (omit a), 18-20, 21 (skip d), 22 [Note: for functional group problems, skip the "cyclic" designation!]	27, 28, 29 (we will see this is crucial to the performance of all proteins!), 30, 31, 33-35, 38-40, 41 (skip c), 42, 44
3	1a, 2a, 3, 4a-e, 5, 6a,b, 7a,b, 9a, 11- 13, 15b-d, 16, 17a,b, 18-21, 25-29	33, 34 (omit c and d), 35 (omit b), 37 (omit e,g,h), 38, 39, 40b, 42, 43a,b, 44, 46
4	1a-c, 2, 3, 4a, 9a, 11-13, 15, 16, 18, 19a-d, 24, 25, 28-32.	35-39, 41, 42a, 43, 44, 46 (skip d) (Be Sure to do 46, very important)
5	2 (label as chiral or achiral. If chiral, also draw the enantiomer.), 3 (star chiral C's, identify each chiral molecule, and be able to draw the enantiomers.), 4, 5 (assign as chiral or achiral), 6 [skip f,g. For all others, give the (R)/(S) designations.], 14, 20a-e, 21 (skip f), 22, 23c	26a,c,d,j-p , 27, 30d, f-h 31a, f-i, 36
6	1, 2c,e,f, 3 (parts 1 and 3, don't classify B or name k), 6, 7 (the density of chloroform is 1.50), 8a, 10 S <sub>N</sub> 2 Reactions: 11-13, 14a,b,d,e, 15(skip b,g), 16, 18 (skip neopentyl bromide. And, substitution is more important than leaving group), 19a,b, 20(skip c,e,f), 21 [(the catch here is to understand why inversion can occur if (S) goes to (S)] S <sub>N</sub> 1 Reactions: 22, 23, 24, 25, 26 (skip the mechanisms, but note how rearrangement forms a more stable cation), 27, 29 (very interesting. Probably not test fodder.) Elimination reactions: 30, 31, 32, 33b-d, 34-39, 40	42a,c-e, 43a-c,e,f, 44**, 45("solvolysis" is substitution by solvent, and is always S <sub>N</sub> 1), 46, 48-54, 56, 59-61

Ch     In Chapter Problems

7     1 (for b, counting geometric isomers, I count 14 possible alkene isomers and 15 possible cyclic isomers! The answer book only shows a few of the possibilities.), 4, 5a,b,c,f,g,h, 6a,d,e, 7a,c,e(name is ambiguous), 8a,c,e, 10a-d (more stable only. Skip the part about how much difference in energy), 12a,c, 13, 16, 17, 18, 19, 24, 25, 27-29 (note: in 28a, 29c,d cation rearrangements occur. I won't ask for mechanisms with cation rearrangement on your test, but a simple elimination of H<sub>2</sub>O such as 29b or c is extremely likely.)

8     1-4, 6, 8-11, 13-21(look at answer to e, just for interest sake), 22 (for b, book answer is poor. Should use a hindered base), 23, 24, 29, 30 (mech for ring-opening only), 32b,d, 33, 34b,d,f, 35 (d,l means racemic mix of chiral products), 36, 37

10    1, 5d, 6, 8, 10, 12a,b,d, 13-16, 17 (esters only), 18-20, 22-26

11    1a,b,d, 2, 3, 5 (skip KMnO<sub>4</sub>), 6, 9, 10, 11, 12a, 13, 14, 22, 23, 27a,b, 33, 34, 35, 36, 37, 38

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7     1 (for b, counting geometric isomers, I count 14 possible alkene isomers and 15 possible cyclic isomers! The answer book only shows a few of the possibilities.), 4, 5a,b,c,f,g,h, 6a,d,e, 7a,c,e(name is ambiguous), 8a,c,e, 10a-d (more stable only. Skip the part about how much difference in energy), 12a,c, 13, 16, 17, 18, 19, 24, 25, 27-29 (note: in 28a, 29c,d cation rearrangements occur. I won't ask for mechanisms with cation rearrangement on your test, but a simple elimination of H<sub>2</sub>O such as 29b or c is extremely likely.)

End of Chapter Problems

31, 32a,b,d, 33, 34 (for part c: how many rings does it have?), 36a-c, 38 (try to predict the major product. For test purposes I usually wouldn't want the minors), 39a,b,d (the point is to predict the major product), 44, 45

47 (good practice for "predict the product" reactions.), 48a, b, c,e,f 49a,b,c,d,e,f,h, 50a-l, 55, 58-61, 68

31, 33a-d, 34b,c, 35a,c, 36b,c, 37 (review from chapter 8), 38a-l, 39, 40, 42, 43

40 (do the bromides only), 41 (skip g), 42, 43, 44, 48a, b, c, f, g, h, 49, 50, 52, 53, 56

47 (good practice for "predict the product" reactions.), 48a, b, c,e,f 49a,b,c,d,e,f,h, 50a-l, 55, 58-61, 68

31, 32a,b,d, 33, 34 (for part c: how many rings does it have?), 36a-c, 38 (try to predict the major product. For test purposes I usually wouldn't want the minors), 39a,b,d (the point is to predict the major product), 44, 45

Chemistry 350, Jasperse, Fall 2010			Reading
Class	Date	Topic	Assignment
1	Aug. 23	Intro. Octet Rule, Lewis Structure, Electroneg, Polarity, Formal Charge	1.1-1.6
2	Aug. 25	Resonance; Structural Formulas; Acids/Bases, Electrophiles/Nucleophiles	1.7-1.14
3	Aug. 27	Orbitals, $\pi$ -Bonds, Hybridization + Shape; Drawing 3-D Shapes	2.1-2.6
	Aug. 30	Bond Rotation, Isomerism, Polarity, Intermolecular Forces, Solubility	2.7-2.11
4	Sept. 1	Classification of Organic Compounds. The Functional Groups.	2.12-2.14
5	Sept. 3	Classification, Formulas, Physical Properties, Nomenclature of Alkanes	3.1-3.5
6	Sept. 6	Labor Day Holiday	No Class
7	Sept. 8	Conformations and Stability of Acyclic Alkanes and Cycloalkanes	3.6-3.12
8	Sept. 10	Conformations and Stability of Cyclohexanes	3.13-3.16
9	Sept. 13	Alkane Chlorination. Factors to Think About in a Chemical Reaction.	4.1-4.9
10	Sept. 15	Transition States, Multistep Reactions, Halogenation of Higher Alkanes.	4.10-4.14
11	Sept. 17	Test 1. Chapters 1-3.	Test
12	Sept. 20	Reactive Intermediates (Radicals, Cations, Anions)	4.15-4.16
13	Sept. 22	Chirality, R/S Classification of Chiral Carbons.	5.1-5.3
14	Sept. 24	Miscellaneous Stereochemistry	5.4-5.9
15	Sept. 27	Diastereomers; More than One Chiral Carbon	5.11-5.16
16	Sept. 29	Catchup	Catchup
17	Oct. 1	Nomenclature, Structure, Properties, Reactivity of Alkyl Halides.	6.1-6.7
		Skip 5.10	
18	Oct. 4	The Sn2 Substitution Reaction.	6.8-6.12
19	Oct. 6	The Sn1 Substitution Reaction.	6.13-6.16
20	Oct. 8	The E1 and E2 Elimination Reactions. Substitution vs. Elimination?	6.17-6.21
	Oct. 11	Fall Breather, NO CLASS	No Class
21	Oct. 13	Catchup	Catchup
22	Oct. 15	Alkenes: Structure, Nomenclature, Isomers.	7.1-7.6
23	Oct. 18	Alkene Stability; Synthesis.	7.7-7.10
24	Oct. 20	Test 2. Chapters 4-6	Test
25	Oct. 22	Synthesis of Alkenes; Classifying/Recognizing Reaction Mechanisms; Alkenes	7.10-8.2
		Skip 7.11	
26	Oct. 25	Addition of H-Cl, H-Br, and H-OH to Alkenes.	8.1-8.5
27	Oct. 27	Oxymercuration/Dermercuration; Hydroboration/Oxidation; Hydrogenation	8.5-8.7,8-10
28	Oct. 29	Addition of Halogens, Formation of Halohydrins; Epoxidation	8.8-8.9
		Skip 8.11	
29	Nov. 1	Oxidation Reactions of Alkenes	8.12-8.16
30	Nov. 3	Catchup; Practice Problems	Catchup
31	Nov. 5	Structure, Nomenclature, Properties, Weak Acidity of Alcohols	10.1-10.6
		Skip 8.16	
32	Nov. 8	Synthesis of Alcohols; Organometallic Reactions.	10.7-10.9
33	Nov. 10	Test 3. Chapters 7,8	Test
34	Nov. 12	Side Reactions; Reduction of Carbonyl Compounds	10.10-10.11
		Skip 10.12	
35	Nov. 15	Catchup; Practice Problems	Catchup
36	Nov. 17	Oxidation of Alcohols	11.1-11.3
37	Nov. 19	Conversion of Alcohols to Tosylates or Halides; Uses of Tosylates and Halides	11.5-11.9
38	Nov. 22	Miscellaneous; Multistep Synthesis	11.10, 11.14
39	Nov. 24	Thanksgiving Break	No class
40	Nov. 26	Thanksgiving Break	No class
41	Nov. 29	Catchup, Multistep Synthesis Problems	Catchup
42	Dec. 1	Review for Test 4	---
43	Dec. 3	Test #4 Covering Chapters 10,11	Test
		Skip 11.4, 11.11-13	
44	Dec. 6	Practice Test for Final, As Much as Time Permits	
	Dec. 9	Final Exam. 9:00 THURSDAY	Final Exam

Tentative  
Letter  
Grades:  
A: 90%  
B: 78%  
C: 66%  
D: 54%