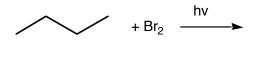
Alkane Brominations:

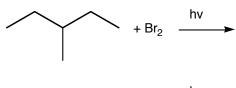
Skills:

- 1. Identify all possible monosubstituted products
- 2. Identify the Major Product
 - Consider all possible radicals. The carbon that gives the most stable radical will be the carbon that gets brominated preferentially.
 - This is true because the rate determining step is the step in which a hydrogen is abstracted and a carbon radical is formed.
 - Thus, according to the product stability/reactivity principle, the pathway leading via the best carbon radical is the preferred path.
- 3. Write the mechanism for chain propagation (with detailed arrows)
- 1. Do all three things for:

Mechanism

2. Identify the Major Product for each of the following:





$$+ Br_2 \rightarrow$$