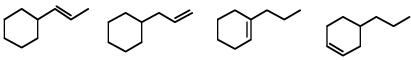
- Ch. 7 Structure and Synthesis of Alkenes
- Ch. 8 Reactions of Alkene

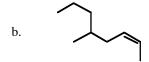
The number of points per problem is indicated in parentheses following each problem.

1. Rank the following alkenes in order of stability, 1 being most stable, 4 being least stable. (4)



2. Determine the number of elements of unsaturation for C₅H₇ClO. (3)

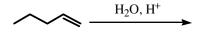
- 3. Give the proper IUPAC name or the structure for the following compounds. (4 points each)
- a. (E)-2-chloro-3-methyl-2-pentene



4. Rank the reactivity of the following alcohols towards HBr, 1 being the fastest reactant, 3 being the slowest reactant. (3 points)

1

5. Predict the <u>major</u> product for the following reactions. You needn't bother to show any side products or minor products. Pay careful attention to orientation. (3 points each)



$$H_2SO_4, \Delta$$

6. Predict the <u>major</u> product in each of the following reactions. Pay careful attention to stereochemistry! (3 points each)

$$\longrightarrow$$
 Br₂

$$OsO_4, H_2O_2$$

7. Fill in the starting reactant. (4 points each)

CH₃CO₃H H OH H OH

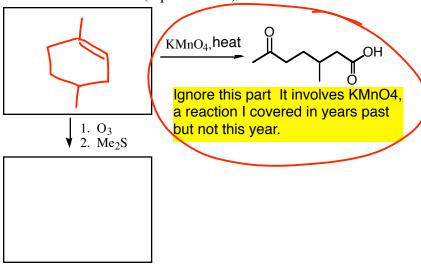
8. Provide the major product of the following reaction sequences. (4 points each)

 $\begin{array}{c}
\text{OH} \\
 & \underline{\text{1. H}_2\text{SO}_4, \text{heat}} \\
\text{2. Br}_2
\end{array}$

Br 1. NEt₃, heat 2. HBr, peroxides
3. NaOCH₃

- 9. What is a possible structure for a molecule A given the following: (6 points)
- a. is has the formula C_6H_{10}
- b. it reacts with H_2/Pt to give a product with formula $C_6H_{12}\,$
- c. upon ozonolysis (O₃; Me₂S) it gives two products, CH₂=O and a product C₅H₈O.

10. Fill in the boxes. (6 points total)



11. Provide reagents to accomplish the following transformations. (6 points each)

$$\longrightarrow$$
 OH

12. Draw the mechanisms for the following reactions. Be sure to draw all intermediates, and try to correctly draw "electron-movement" arrows. (8 points for the first, 6 points for the second)

$$\begin{array}{c}
\text{OH} \\
\frac{\text{H}_2\text{SO}_4,}{\text{heat}}
\end{array}$$

$$HBr$$
 CH_2
 HBr
 CH_3

13. Draw as many isomers as you can for alkenes with formula C_5H_{10} . (8 points. 2 points off for each duplicate or each possible isomer not drawn.)