Ch 18 Ketones and Aldehydes Ch 22 Additions and Conensations of Enols and Enolate Ions

1. Nomenclature. Provide the structure or the name for the following. If stereochemistry is a factor, do not neglect it. (6 pt)

## 3-isopropylbenzaldehyde

optically active

2. Rank the following, with 1 being highest, or most. (6 pt)

Equilibrium concentration of enol

Reactivity toward MeMgBr

Acidity

3. Draw the products for the following reactions (3 pt each)

$$\begin{array}{c}
\bullet \\
\bullet \\
\bullet
\end{array}$$

$$\begin{array}{c}
H_2O, H^+
\end{array}$$

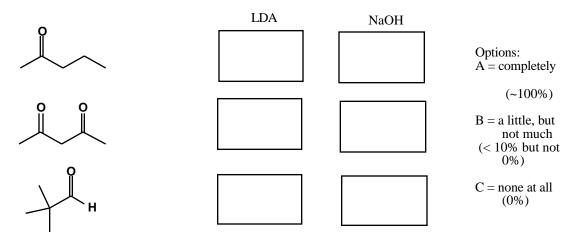
$$\mathbf{CH_3}\mathbf{-C}\mathbf{\equiv C}\mathbf{-H} \quad \frac{\mathbf{Hg^{2+},\,H_2O}}{\mathbf{H_2SO_4}}$$

4. Draw the products for the following multistep reactions. (3 pt each)

OMe 
$$\frac{1. \text{ H}^+, \text{ HO}}{2. 2 \text{ MeMgBr}}$$

$$3. \text{ H}_2\text{O}, \text{H}^+, \text{ heat}$$

5. For the following chemicals, describe the extent to which each would be deprotonated by LDA (LiN-iPr<sub>2</sub>) or by NaOH at equilibrium. Fill in all 6 boxes. Options are complete deprotonation (A), a little deprotonation (B), and no deprotonation (C). (6 pt)



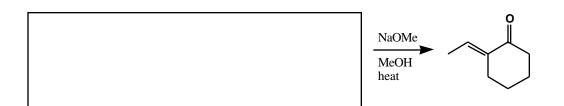
- 6. Suggest a plausible structure consistent with the following information. (5 pt)
- a. It reacts positively with 2,4-dinitrophenylhydrazine.
- b. It reacts positively with NaOH/I<sub>2</sub>, the iodoform test
- c. It does not react with Tollen's reagent  $[Ag(NH_3)_2^+OH^-]$ .
- d. It does not react with Br<sub>2</sub> in dichloromethane solvent.
- e. Chemical formula is C7H12O
- f. It's <sup>13</sup>C spectrum shows 5 carbons (1 singlet, 1 doublet, 2 triplets, and 1 quartet)

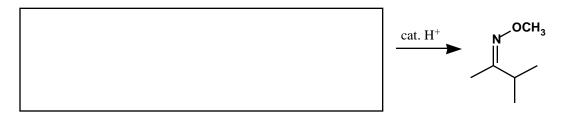
7. Which of the following would <u>not</u> undergo decarboxylation (loss of CO<sub>2</sub>) upon heating? (2 pt)

8. Put in the starting materials from which the following would be made. (3 each)









9. Draw the mechanisms for the following reactions. (4 pt each)

O O OH

NaOH II I

(Note: this one counts as 2 problems, 8 points total)

10. Provide reagents for the following transformations. (4 pt each)

11. Design a synthesis for the following alkene **FROM ALCOHOLS WITH NO MORE THAN 5 CARBONS**. (6 pt)