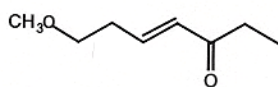


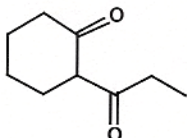
Jasperse
Chem 360
Version 1
Test 3
Answers

JASPERSE CHEM 360 TEST 3 VERSION 1
Ch 18 Ketones and Aldehydes
Ch 22 Additions and Conensations of Enols and Enolate Ions

1. Provide the Name for the Following (6 points)



trans-7-methoxy-4-hepten-3-one
(or E) (E)-7-methoxyhept-4-en-3-one



2-propanoylcyclohexanone

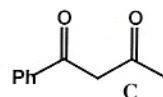
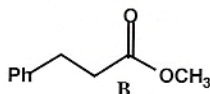
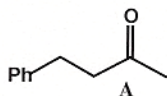
2. Of the following structures,

a. Which will be "completely" (>98%) deprotonated by LDA (LiN-iPr₂)? (2 points)

A, B, C

b. Which will be "completely" (>98%) deprotonated by NaOH? (2 points)

C only



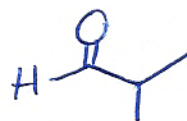
3. An unknown X has formula C₅H₈O. It gives 1) an orange precipitate upon treatment with 2,4-dinitrophenylhydrazine (2,4-DNP) and it gives 2) a silver mirror upon treatment with Tollen's reagent [Ag(NH₃)₂⁺OH⁻]. 3) It does not react with Br₂ in dichloromethane solvent. 4) Included in the ¹H NMR (incomplete) is a 6H doublet at 1.2 ppm. What is X? (4 points)

EU=1

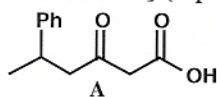
C=O

aldehyde

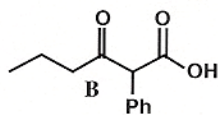
NMR ⇒ isopropyl



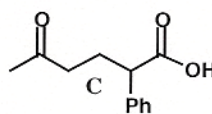
4. Rank the rate of decarboxylation (loss of CO₂) for the following molecules upon heating, with 1 being highest, 2 being next, and 3 being not at all. [Hint: Two out of the three will react, one will not, so you should be able to identify the unreactive isomer. To compare the reactivity of the two reactive isomers, the phenyl substituent impacts the relative stabilities in the key step of the mechanism.] (2 points)



2



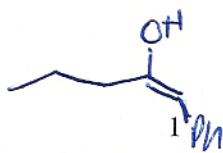
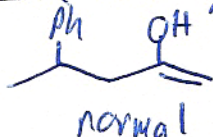
1



3

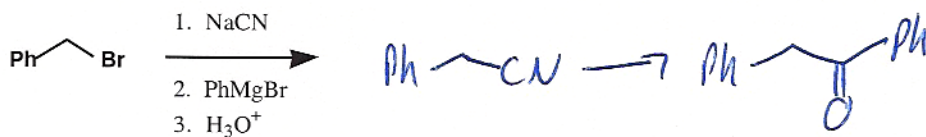
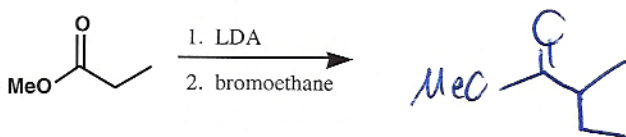
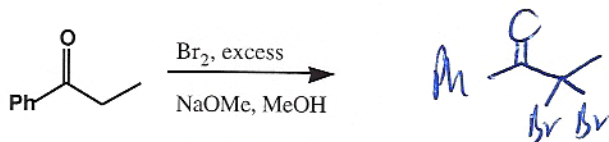
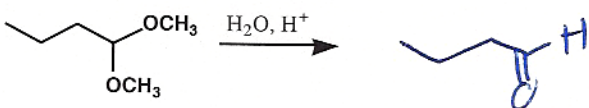
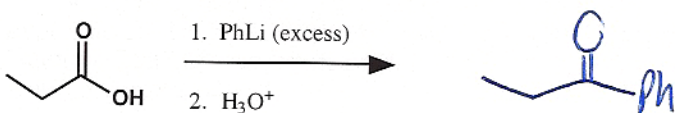
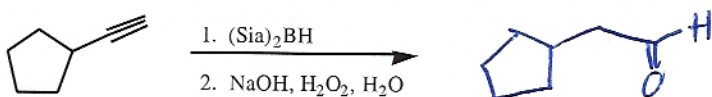
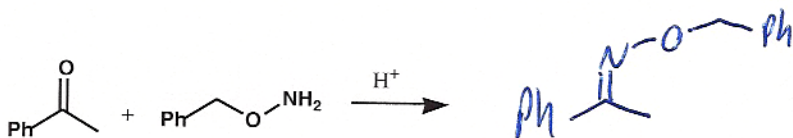
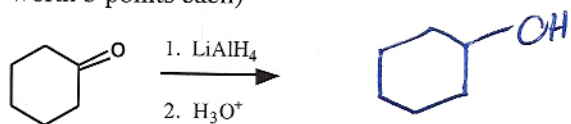
not 1,3 relationship

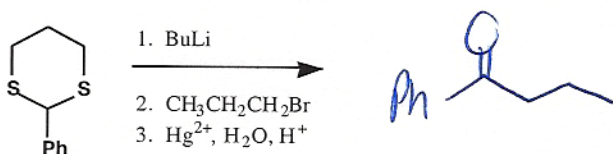
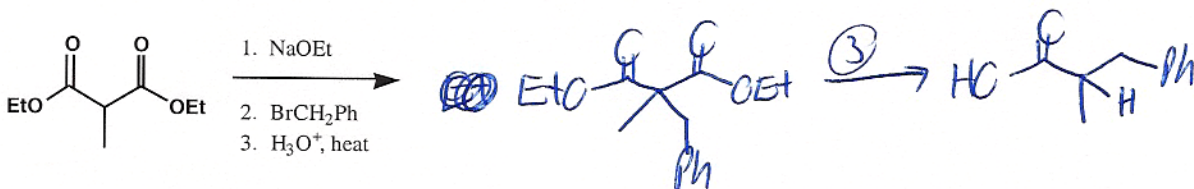
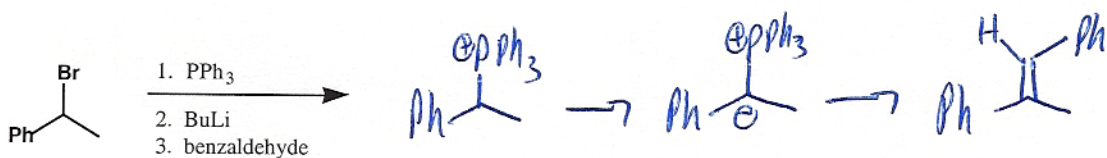
Enol stability:



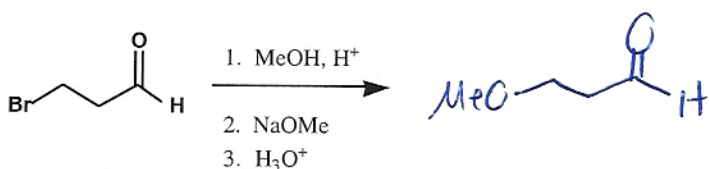
stabilized
by conjugation

5. Synthesis Reactions. Draw the feature product of the following reactions (need not show any byproducts). NOTE: In every case, the product should be a stable, isolable **product**; an "intermediate" structure will not receive full credit. (2 or 3 points each; 1st 7 worth 2 points; last 5 worth 3 points each)



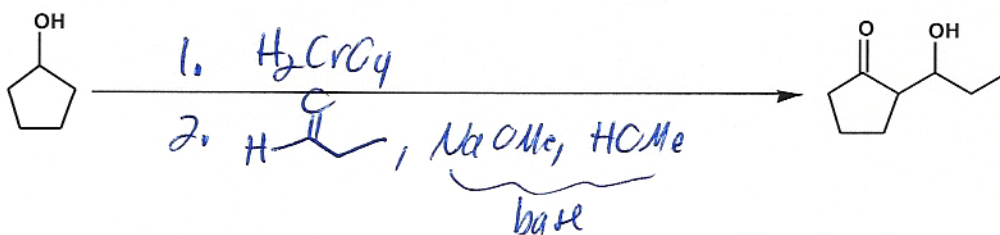
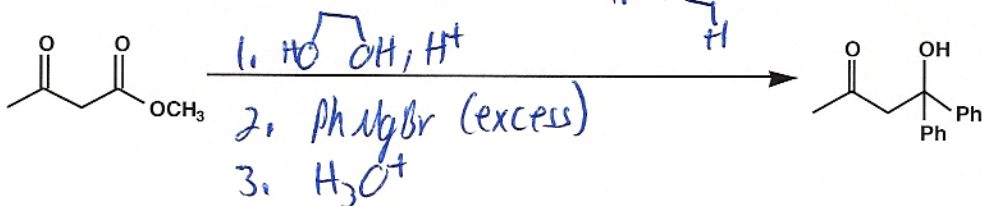
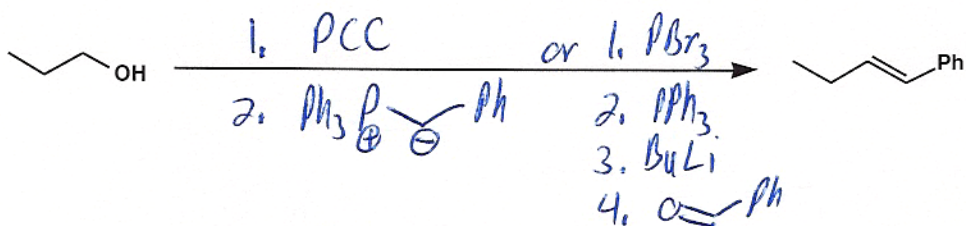


Not responsible for this problem, I covered less this year

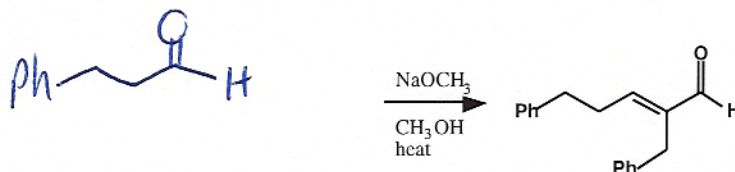
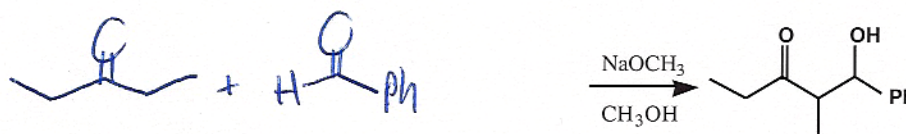
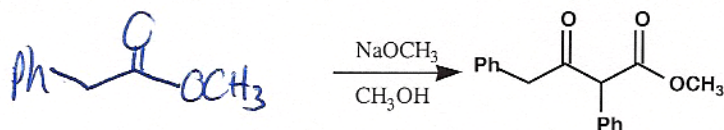
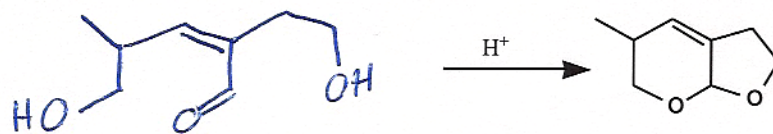


Protect-S_N2-deprotect

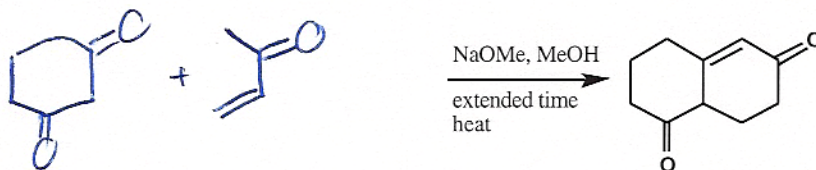
6. Provide Reagents for the Following Transformations: (4 points each)



7. Put in the starting materials from which the following structures would be produced. Depending on the product, the appropriate starting material may be either a single molecule, two of the same molecule, or two different molecules. For the last problem, you are required to start from two separate molecules. (2 points each)

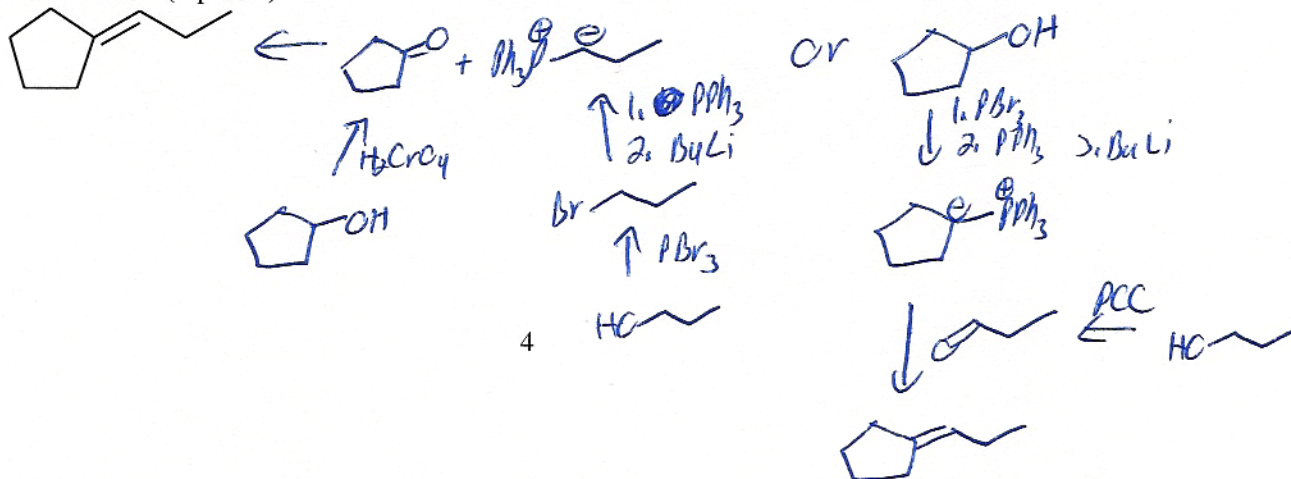


Note: The Starting Materials are two Separate Molecules

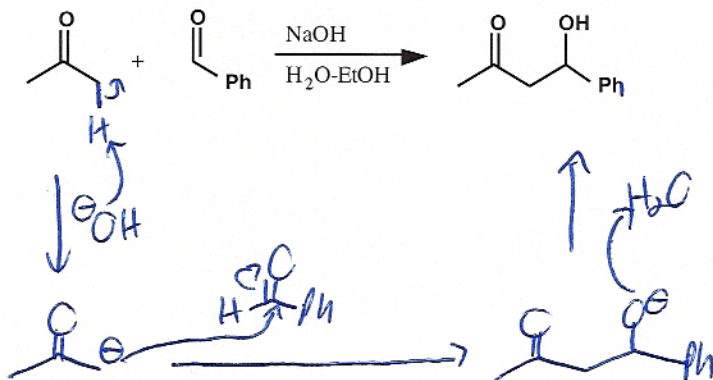
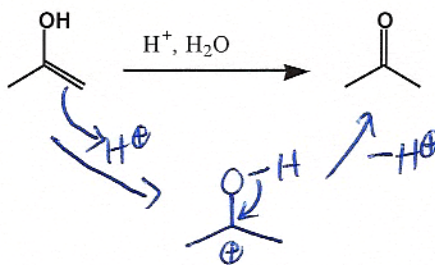
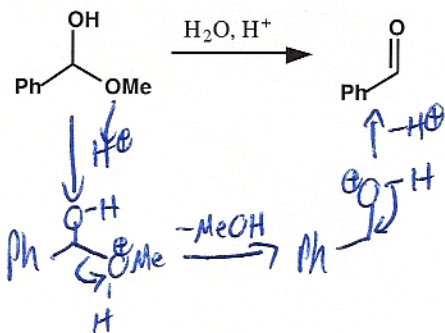
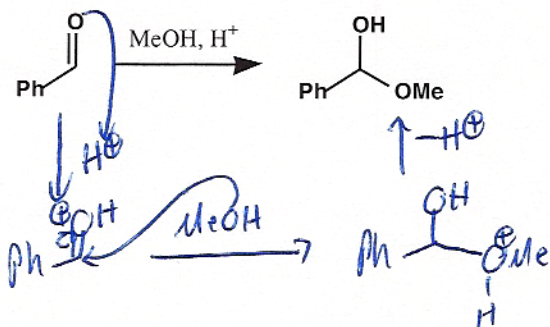
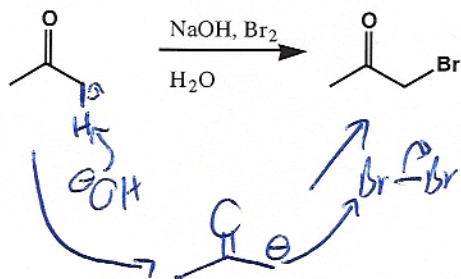


Not responsible, I covered less this year.

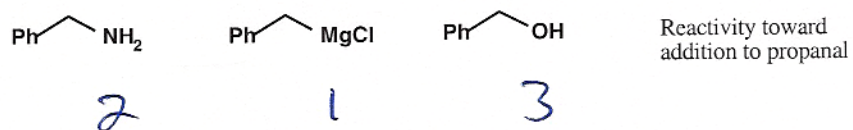
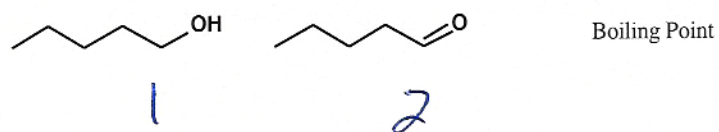
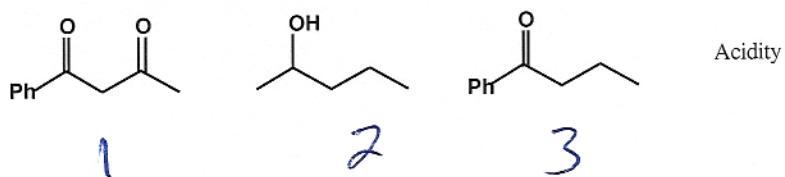
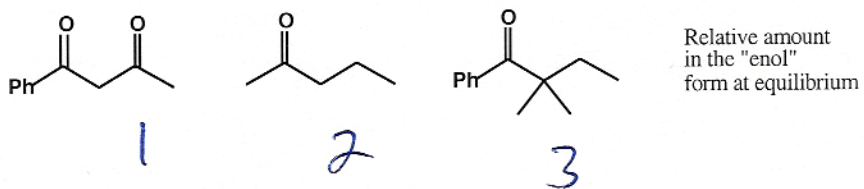
8. Design a synthesis for the following alkene, FROM ALCOHOLS WITH NO MORE THAN 5 CARBONS. (4 points)



9. Provide Mechanisms for the Following Transformations. [Note: Some of these do not represent "clean" reactions; the product shown might go on to further reactions, or the reaction might be reversible, or the product might not be isolable. But that shouldn't prevent you from drawing the mechanism for the transformation indicated!] (3 points each)



10. Rank the following, with 1 being highest, or most. (2 points each)



Not responsible for 2 vs 3, why the amine is more reactive than the alcohol.

