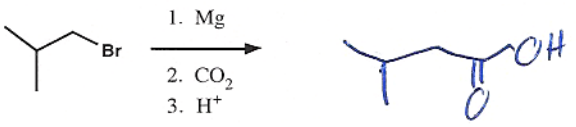
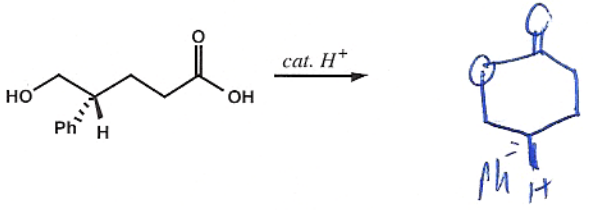
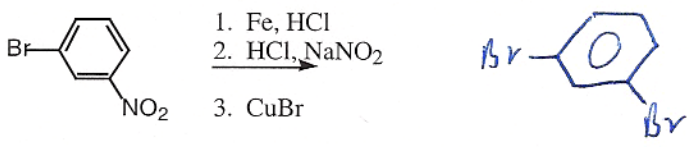
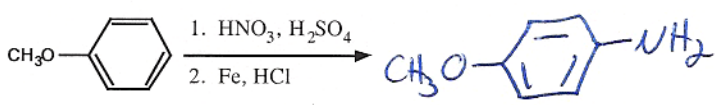
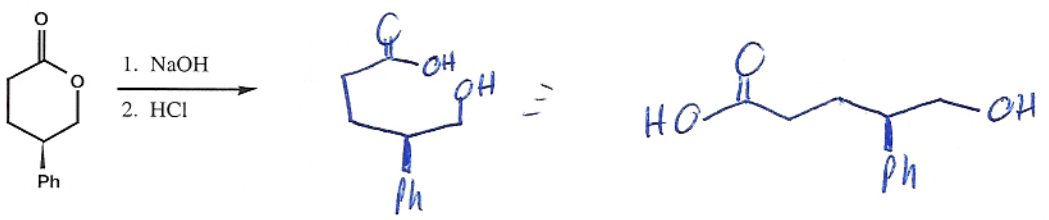
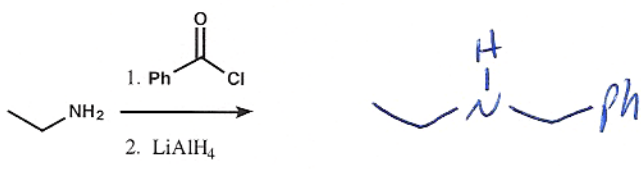
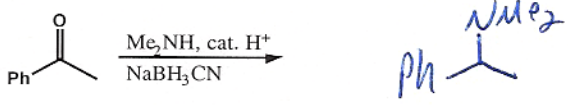
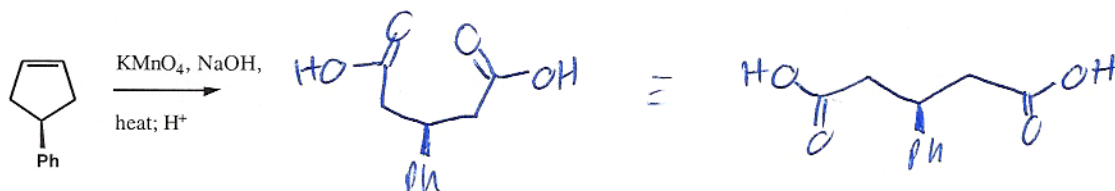
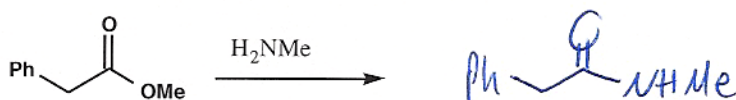
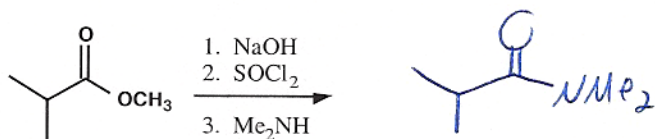


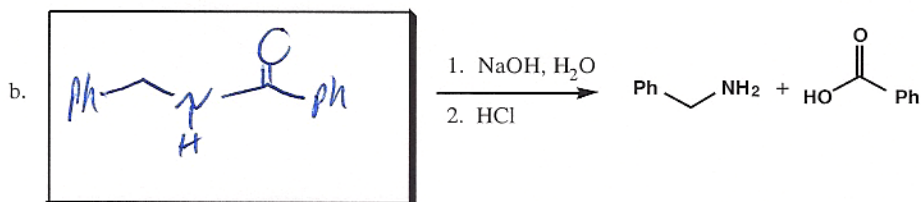
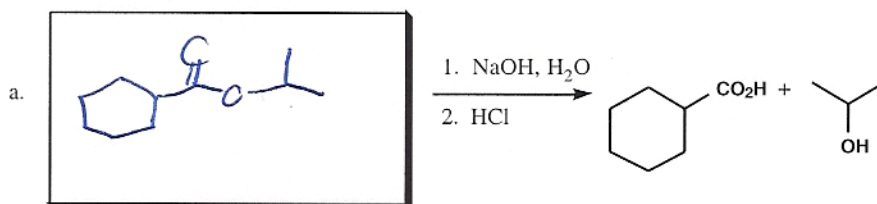
Jasperse  
Chem 360  
Test 4  
Version 2  
Answers

1. Synthesis Reactions. Draw the feature product of the following reactions (need not show any byproducts). (22 points, 2 points each)

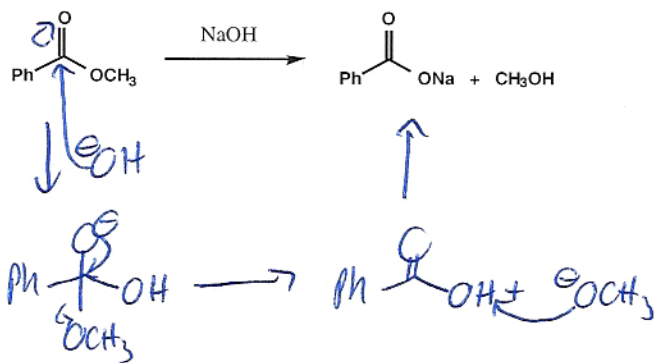
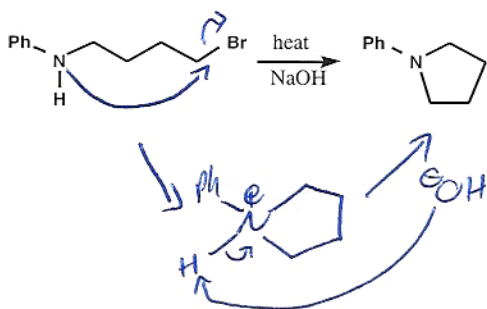
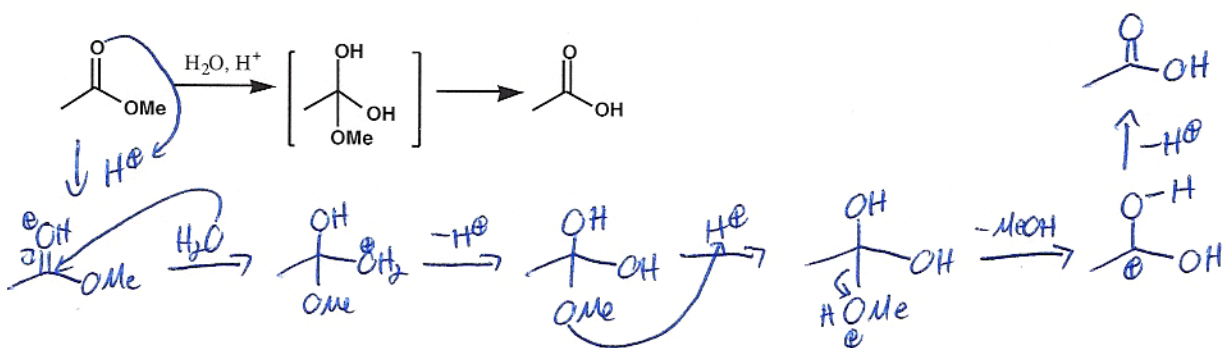
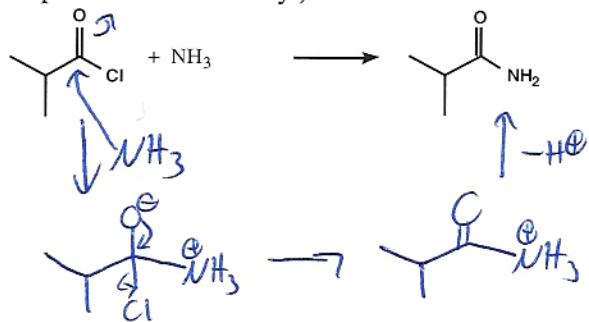




2. Hydrolysis Reactions. Draw the starting materials for the following hydrolysis reactions. (4 points)

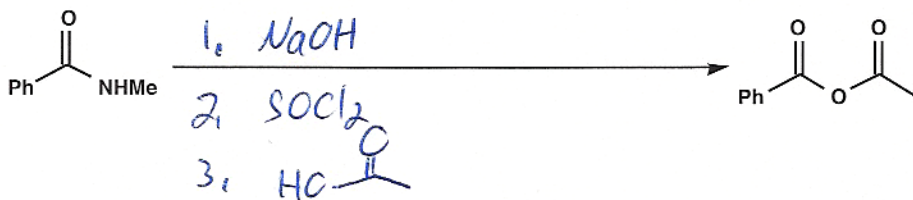
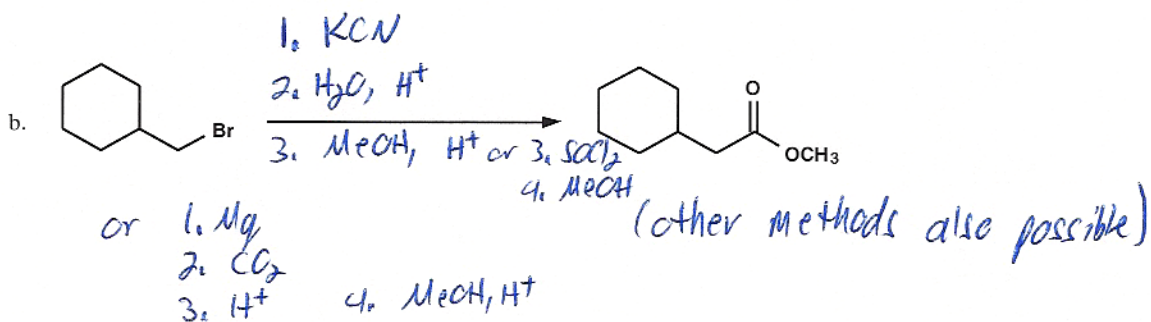
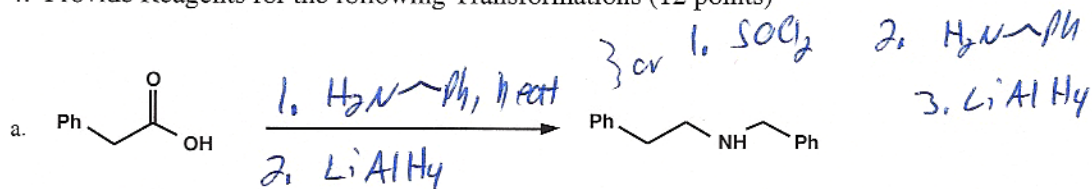


3. Draw the Mechanisms for the following reactions. (16 points total. Some are relatively trivial, so point values will vary.)



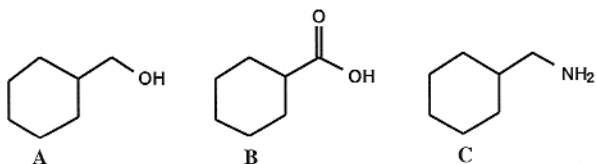
Na<sup>+</sup> just a spectator, ignore.

4. Provide Reagents for the following Transformations (12 points)



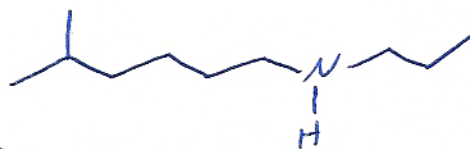
5. Which (if any) after being dissolved in diethyl ether, will: (6 points. Note: The answers may be none or more than one, you tell me!)

- a) Extract into NaOH/H<sub>2</sub>O? **b**  
 b) Extract into HCl/H<sub>2</sub>O? **c**  
 c) Extract into water? **none**

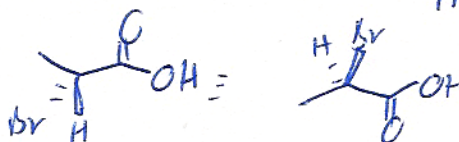


6. Nomenclature. Provide Either the Name or the Structure for the Following Chemicals. (8 points)

a. N-propyl-5-methyl-1-hexanamine  
N-propyl-5-methylhexan-1-amine



b. (R)-2-bromopropanoic acid



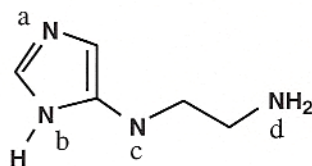
d.

3-hydroxy pentanoic acid

e.

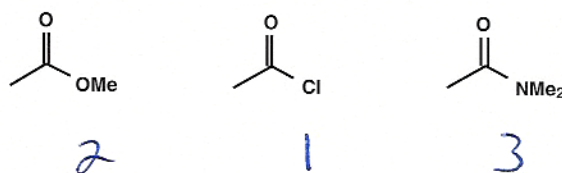
N,N-diethyl-1-propanamine  
N,N-diethylpropan-1-amine

7. For each nitrogen a-d, identify the hybridization of the nitrogen atom, and identify the hybridization of the nitrogen lone pair. (6 points, 2 points off for 1st error, 1 for each additional)

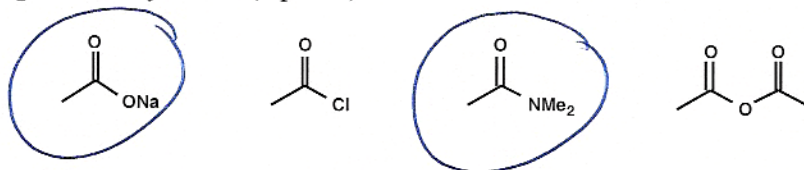


Nitrogen Atom	Hybridization of the Nitrogen Atom	Hybridization of the Nitrogen Lone Pair
a	$sp^2$	$sp^2$
b	$sp^2$	p
c	$sp^2$	p
d	$sp^3$	$sp^3$

8. Rank the following according to their reactivity toward NaOH/H<sub>2</sub>O hydrolysis, from 1 (most) to 3 (least). (2 points)



9. Circle the compounds, if any, (may be none, one, or more than one) that would not react with methanol to give a methyl ester: (4 points)



10. Rank the acidity of the following, 1 being most acidic, 3 being least (9 points)

- a.  $\text{CH}_3\text{NH}_3^+\text{Cl}^-$       benzoic acid      water  
2                                  1                                  3
- b.  $\text{CH}_3\text{CO}_2\text{H}$                $\text{CH}_3\text{CH}_2\text{OH}$                $\text{CH}_3\text{CH}_2\text{NH}_2$   
1                                  2                                  3
- c. water                      p-nitrobenzoic acid              p-methylbenzoic acid  
3                                  1                                  2

11. Rank the basicity of the following, 1 being most basic, 3 being least (9 points)

- a.  $\text{PhNH}_2$                        $(\text{CH}_3)_3\text{N}$                        $\text{CH}_3\text{NH}_2$   
3                                  1                                  2
- b.  $\text{Me}_2\text{NH}$                        $\text{CH}_3\text{CO}_2\text{Na}$                        $\text{H}_2\text{O}$   
1                                  2                                  3
- c.  $\text{NaOH}$                        $\text{CH}_3\text{MgBr}$                       pyridine  
2                                  1                                  3