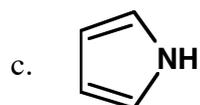


JASPERSE CHEM 350 TEST 4 VERSION 2
 Conjugation, Aromatic Compounds, Reactions of Aromatic Compounds

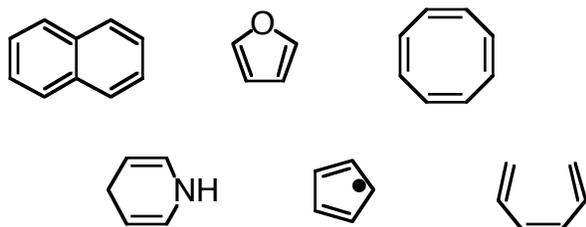
1. Provide the Name of Structure for the following. (7 points)

a. 3-nitroaniline

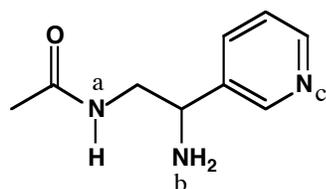
b. o-isopropyltoluene



2. Circle the aromatic molecules. (7 points)



3. The molecule has 3 different nitrogens. For each of them, classify the hybridization of the nitrogen atom, the hybridization of the nitrogen lone pair, and whether the basicity of the nitrogen is "normal" or "low". (6 points)



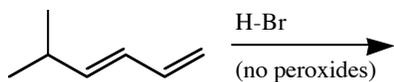
Nitrogen Hybridization	Lone-Pair Hybridization	Nitrogen Basicity
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N^a

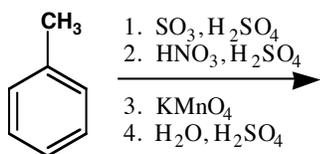
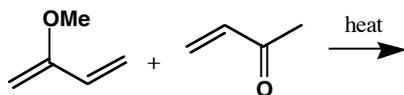
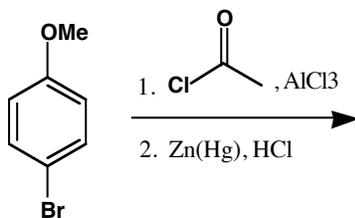
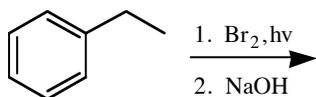
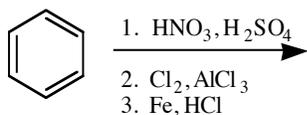
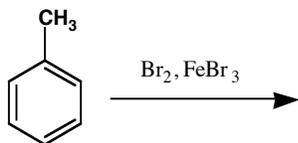
N^b

N^c

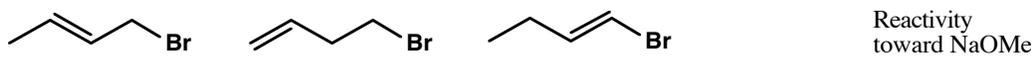
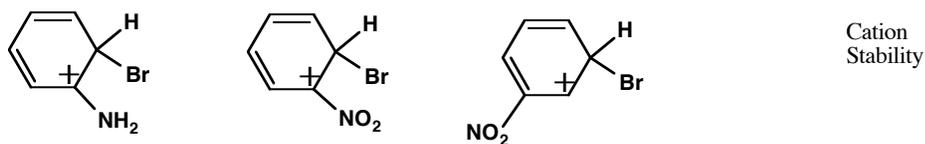
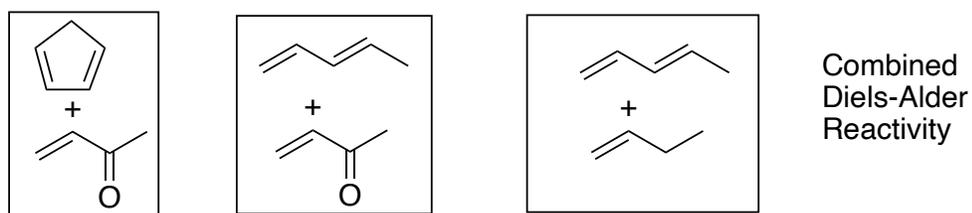
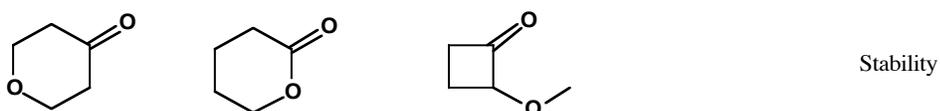
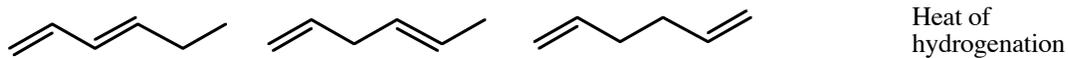
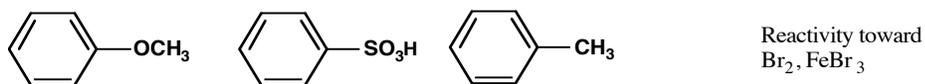
4. Draw the major products of the following reaction (4 points).



5. Draw the Major Product of the Following Reactions. Note: I want one major product in each case. (3 points each)



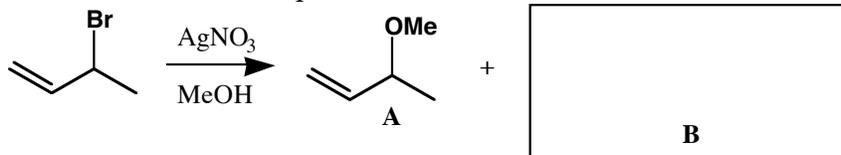
6. Rank the following, with 1 being highest/most. (2 points each)



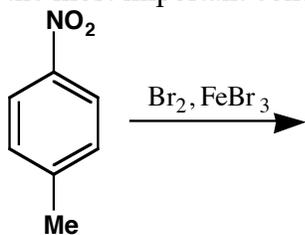
8. Treatment of an alkyl halide with methanolic AgNO_3 often promotes ionization, via the following:



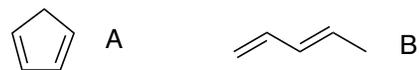
When 3-bromo-1-butene undergoes this reaction, two isomeric products **A** and **B** are formed. Draw the structure for product **B** and the detailed mechanism for formation of product **B**. (7 pts)



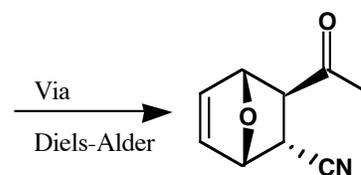
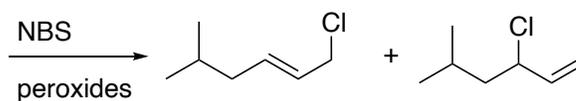
9. Draw the product for the following reaction and draw the mechanism for its formation. Identify the slow step. Draw all the resonance structures for the cation intermediate and circle the most important contributor. (7 points)



10. (3 pt) When comparing cyclopentadiene (**A**) versus 1,3-pentadiene ($\text{CH}_2=\text{CH}-\text{CH}=\text{CH}-\text{CH}_3$, **B**), one is much more acidic. Circle the more acidic chemical, and explain why it is much more acidic.



11. Draw the Reactants for the Following Reactions (6 points)



12. Provide reagents for the following transformations. (5 points each)

